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### Flinn Chemtopic Labs Restoring Balance Answer

Restoring Balance: LeChâtelier's Principle and Equilibrium—ChemTopic™ Lab Activity. Chemical equilibrium is a true balancing act. With the Restoring Balance: LeChâtelier's Principle and Equilibrium—ChemTopic™ Lab Activity, observe the effects of reaction conditions, such as concentration and temperature, on equilibrium and visualize how balance can be restored based on LeChâtelier's Principle.

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### Flinn Chemtopic Labs Organic Compounds Answers

Restoring Balance Lab. 1. The solubility of iodine increases because the position of equilibrium has shifted towards the triiodide side of the reaction. Addition of the reactants causes an increase in the product which is directly related to the solubility of iodine in water in this case. 2.

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